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# Structural analysis of actinidin and a comparison of cadmium and sulfur anomalous signals from actinidin crystals measured using in-house copperand chromium-anode X-ray sources 

The structure of the 24 kDa cysteine protease saru-actinidin from the fruit of Actinidia arguta Planch. (sarunashi) was determined by the cadmium/sulfur-SAD method with X-ray diffraction data collected using in-house $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} \mathrm{K} \alpha$ radiation. The anomalous scatterers included nine sulfurs and several cadmium ions from the crystallization solution. The high quality of the diffraction data, the use of chromiumanode X-ray radiation and the substantial anomalous signal allowed structure determination and automated model building despite both a low solvent content ( $<40 \%$ ) and low data multiplicity. The amino-acid sequence of saru-actinidin was deduced from the cDNA and was modified based on experimental electron-density maps at $1.5 \AA$ resolution. The active site of saru-actinidin is occupied by a cadmium ion and the active-site cysteine is found to be in an unmodified, cysteine sulfenic acid or cysteine sulfinic acid form. The cadmium sites, coordination geometries and polygonal water structures on the protein surface have also been extensively analyzed. An analysis and comparison of the sulfur/cadmium anomalous signals at the $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} K \alpha$ wavelengths was carried out. It is proposed that the inclusion of cadmium salts in crystallization solutions coupled with chromium-anode radiation can provide a convenient route for structure determination.

## 1. Introduction

Actinidia arguta Planch. (sanusashi; also known as hardy kiwi) produces a small kiwi-like fruit from which a novel cysteine protease, saru-actinidin, has been isolated. Saru-actinidin shares $\sim 50-81 \%$ sequence identity with previously structurally determined plant cysteine proteases, including actinidin from A. chinensis (chins-actinidin; Baker \& Dodson, 1980), papain, caricain, chymopain and glycyl endopeptidases from Carica papaya (Pickersgill et al., 1991, 1992; Maes et al., 1996; Groves et al., 1996), ervatamins A, B and C from Tabernaemontana divaricata (Biswas et al., 2003; Thakurta et al., 2004), ginger protease II (GPII) from Zingiber officinale (Choi et al., 1999), mexicain from Jacaratia mexicana and endoprotease B isoform 2 (EP-B2) from Hordeum vulgare (Bethune et al., 2006). Unprotected active-site cysteine residues can undergo various covalent modifications which may regulate protein activity (Reddie \& Carroll, 2008). Cysteine can form three different species by reacting with oxygen: cysteine sulfenic $(-\mathrm{SOH})$, cysteine sulfinic $\left(-\mathrm{SO}_{2} \mathrm{H}\right)$ and cysteine sulfonic

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$\left(-\mathrm{SO}_{3} \mathrm{H}\right)$ acids. The cysteine proteases were purified in inactive forms in which the active-site cysteines were protected by adducts (Choi et al., 1999; Maes et al., 1996; Biswas et al., 2003; Thakurta et al., 2004; Ghosh et al., 2008). Unprotected forms and modified cysteine sulfinic acid and cysteine sulfonic acid forms have also been reported (Kamphuis et al., 1984; Baker \& Dodson, 1980).

The use of chromium-anode X-ray radiation $(\lambda=2.29 \AA)$ is attractive for single-wavelength anomalous diffraction (SAD) experiments. For example, in the case of sulfur the anomalous scattering signal is more than doubled $\left(f^{\prime \prime}=1.14\right)$ compared with conventional $\mathrm{Cu} K \alpha$ X-rays $\left(f^{\prime \prime}=0.56\right)$. Furthermore, naturally bound metals and atoms from crystallization solutions (e.g. $\mathrm{Cl}, \mathrm{P}$ or Cd ) tend to show a significant increase in anomalous scattering with Cr radiation. The anomalous contribution ( $f^{\prime \prime}$ ) of cadmium at the $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} \mathrm{K} \alpha$ wavelengths is $\sim 4.7$ and 8.97 , respectively. Cadmium may replace calcium and is expected to have similar chelation by
acidic protein residues (Naismith et al., 1993; Harrop et al., 1996). It has been shown previously that some divalent cations can induce or improve the crystallization of proteins (Trakhanov et al., 1998; Trakhanov \& Quiocho, 1995). For example, the addition of $\sim 200 \mathrm{~m} M$ cadmium ion improved the diffraction quality of a variety of protein crystals (Yao et al., 1994; Nickitenko et al., 1995; Trakhanov \& Quiocho, 1995; Trakhanov et al., 1998; Liu et al., 2006; Stegmann et al., 2010). In most of these the bound cadmium was located at the molecular interface/protein surface and there was no evidence of significant changes in the three-dimensional structure regardless of cadmium concentration (Yao et al., 1994; Adams \& Jia, 2005; Boraston et al., 2006; Stegmann et al., 2010). So far, few structures have been solved by the cadmium-SAD or SIRAS methods (Harrop et al., 1996; Liu et al., 2006; Boraston et al., 2006; Kaus-Drobek et al., 2007; Kajander et al., 2007; Xu et al., 2008; Grimshaw et al., 2008; van Bueren et al., 2009; Gangelhoff et al., 2009; Stegmann et al., 2010) and in most of


Figure 1
Anomalous difference Fourier maps around S atoms and bound cadmium ions at $4 \sigma$ (magenta). (a) A native crystal grown in the presence of $30 \mathrm{~m} M$ $\mathrm{CdCl}_{2}$ was used to collect data to $1.5 \AA$ resolution using $\mathrm{Cu} K \alpha$ radiation $\left(\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}\right.$ data) , (b) $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data were collected to $1.9 \AA$ resolution, (c) $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ data were collected to $2.2 \AA$ resolution and (d) $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data were collected to $2.2 \AA$ resolution. Met and Cys residues are shown as green sticks and bound cadmium ions are highlighted as brown spheres. Two closely related cadmium ions were located in the CD4 site in all of the data sets apart from $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$.
those cases the cadmium ions arose from the crystallization solution. Here, we report the structure determination of saruactinidin by the cadmium/sulfur-SAD technique using four different data sets collected at in-house $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} \mathrm{K} \alpha$ radiation sources and the anomalous signals were analyzed. The saru-actinidin crystals were grown in the presence of $30 \mathrm{~m} M \mathrm{CdCl}_{2}$ (part of the crystallization solution) and data sets, namely $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$, were collected at the $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} K \alpha$ wavelengths. In order to compare and utilize the potentially enhanced anomalous signal for SAD phasing, we increased the concentration of $\mathrm{CdCl}_{2}$ from 30 to $300 \mathrm{~m} M$ using the quick cryosoaking method and two further data sets $\left(\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}\right.$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ ) were collected.


(c)

## 2. Materials and methods

### 2.1. Crystallization

The saru-actinidin crystals were grown by the hanging-drop vapour-diffusion method at 288 K . The crystallization droplet consisted of $2 \mu \mathrm{l}$ protein solution ( $15 \mathrm{mg} \mathrm{ml}^{-1}$ in buffer consisting of $20 \mathrm{~m} M$ sodium phosphate pH 6.0 ) and $2 \mu \mathrm{l}$ reservoir solution $\left[30 \mathrm{~m} M \mathrm{CdCl}_{2}\right.$ and $30 \%(\mathrm{v} / \mathrm{v})$ PEG 400 in 0.1 M sodium acetate pH 4.6 ] and was equilibrated against $300 \mu \mathrm{l}$ well solution. Rod-shaped crystals appeared after 3 d and belonged to the orthorhombic space group $P 2_{1} 2_{1} 2_{1}$ with one molecule per asymmetric unit. The Matthews coefficient and solvent content are $1.96 \AA^{3} \mathrm{Da}^{-1}$ and $\sim 37 \%$, respectively.

### 2.2. Data collection and processing

The X-ray diffraction data sets were collected at the High Intensity X-ray Laboratory, Nagoya University at 100 K using $\mathrm{Cu} K \alpha$ radiation $(\lambda=1.54 \AA)$ and $\mathrm{Cr} K \alpha$ radiation ( $\lambda=2.29 \AA$ ) generated by a Rigaku FR-E rotating-anode X-ray generator operated at 45 kV and 45 mA with Osmic mirrors. The diffraction images were recorded using a Rigaku R-AXIS VII detector. The crystals were cryosoaked by adding $20 \%$ glycerol to the mother liquor prior to data collection. Two data sets $\left(\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}\right.$ and $\left.\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}\right)$ were collected using $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} K \alpha$ radiation, respectively. The $\mathrm{CdCl}_{2}$ concentration in the mother liquor was increased from 30 to $300 \mathrm{~m} M$ in the cryosolution and two further data sets ( $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ ) were collected using $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} K \alpha$ radiation, respectively. For each data set, a total of 360 frames were collected using $1^{\circ}$ oscillation steps with 180 s exposure per frame in each case. The crystal-todetector distance was set to $70,100,86$ and 86 mm for

Figure 2
Plots of the Bijvoet ratio $\langle\Delta F\rangle /\langle F\rangle$, measurability $|\Delta I| / \sigma(\Delta I)$ and anomalous signal-to-noise ratio $\langle(\Delta F) / \sigma(\Delta F)\rangle$ as a function of resolution. (a) Experimentally measured Bijvoet ratio together with the theoretical curve. Dotted blue and red lines represent the theoretically expected Bijvoet ratio for $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} K \alpha$ wavelengths for nine sulfurs and six cadmium ions, respectively, calculated using the modified Hendrickson formula. (b) Measurability curve calculated from experimental data using phenix.reflection statistics. It is defined as the fraction of Bijvoet-related intensity differences for which $|\Delta I| / \sigma(\Delta I)>3.0, \min \left[I^{+} / \sigma\left(I^{+}\right), I^{-} / \sigma\left(I^{-}\right)\right]>3.0$ holds. (c) Anomalous signal-to-noise ratio versus resolution calculated using HKL2MAP.

Table 1
Data-collection statistics.
Values in parentheses are for the last shell.

|  | $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ |
| :---: | :---: | :---: | :---: | :---: |
| Wavelength (A) | $1.54[\mathrm{Cu} \mathrm{K} \alpha]$ | $1.54[\mathrm{Cu} \mathrm{K} \alpha]$ | $2.29[\mathrm{Cr} \mathrm{K} \alpha]$ | $2.29[\mathrm{Cr} \mathrm{K} \alpha]$ |
| Crystal-to-detector distance (mm) | 70 | 100 | 86 | 86 |
| Unit-cell parameters ( $\AA$ ) | $a=48.79, b=56.18, c=70.88$ | $a=47.27, b=55.81, c=70.23$ | $a=48.74, b=56.10, c=70.76$ | $a=48.52, b=56.04, c=70.92$ |
| Space group | $P 2.22_{1}{ }_{1}$ | $P 2_{1} 2_{1} 2_{1}$ | $P 2_{1} 2_{1} 2_{1}$ | $P 2_{1} 2_{1} 2_{1}$ |
| Mosaicity ( ${ }^{\circ}$ ) | 0.8 | 1.2 | 1.1 | 0.6 |
| Resolution ( ${ }^{\circ}$ ) | 110-1.50 (1.55-1.50) | 90.0-1.90 (1.97-1.90) | 90.0-2.20 (2.28-2.20) | 110.0-2.20 (2.28-2.20) |
| Unique reflections | 60045 (5964) | 15226 (1494) | 10270 (932) | 10212 (914) |
| Multiplicity | 7.0 (6.6) | 12.8 (13.0) | 11.9 (7.9) | 6.8 (4.3) |
| Completeness (\%) | 99.5 (98.4) | 99.8 (100) | 99.2 (92.3) | 99.3 (94.2) |
| $\langle I / \sigma(I)\rangle$ | 45.40 (9.64) | 41.28 (6.13) | 41.15 (5.11) | 51.25 (33.81) |
| $R_{\text {merge }}$ | 0.038 (0.251) | 0.083 (0.275) | 0.071 (0.216) | 0.046 (0.077) |
| $\langle\Delta F / \sigma(\Delta F)\rangle$ | 2.5 (1.31) | 2.1 (0.62) | 2.8 (0.7) | 5.9 (4.3) |

Table 2
Summary of phasing and model building.

|  | $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ |
| :--- | :--- | :--- | :--- | :--- |
| AutoSol |  |  |  |  |
| Bijvoet pairs | 27805 | 12646 | 8395 | 8631 |
| Mean anomalous signal $(\langle\Delta F\rangle /\langle F\rangle)$ | 0.04 | 0.05 | 0.06 | 0.07 |
| Solvent fraction | 0.39 | 0.36 | 0.38 | 0.38 |
| Bayesian CC | $56.9 \pm 14.4$ | $43.6 \pm 19.8$ | $54.3 \pm 15.3$ | $58.8 \pm 14.1$ |
| FOM | 0.53 | 0.43 | 0.51 | 0.58 |
| Sites used for phasing | 7 | 8 | 5 | 7 |
| Refined sites | 18 | 11 | 16 | 32 |
| Located sulfur sites | 9 | 3 | 9 | 9 |
| AutoBuild |  |  |  |  |
| Residues built | 211 | 205 | 218 | 219 |
| Side chains | 200 | 197 | 198 |  |
| $R_{\text {work }} / R_{\text {free }}$ | $0.21 / 0.23$ | $0.22 / 0.26$ | $0.19 / 0.25$ | $0.19 / 0.22$ |
| Model-map CC | 0.77 | 0.75 | 0.81 |  |

## 3. Results and discussion

### 3.1. Substructure determination of saru-actinidin

Initially, diffraction data were collected to $1.5 \AA$ resolution ( $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ ) using a native crystal and $\mathrm{Cu} K \alpha$ radiation. Saru-actinidin contains 220 amino acids, including seven Cys and two Met residues. From the crystal parameters, it was clear that there was one molecule per asymmetric unit, with a Matthews coefficient of $1.96 \AA^{3} \mathrm{Da}^{-1}$ and a solvent content of $\sim 37 \%$. The mean
the $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}, \mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}, \mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets, respectively. The data sets were processed and scaled with $D E N Z O$ and SCALEPACK (Otwinowski, 1990). The anomalous pairs were merged separately as $I^{+}$and $I^{-}$. The relevant data-collection statistics are summarized in Table 1.

### 2.3. Software packages used

The programs HKL2MAP (Pape \& Schneider, 2004) and PHENIX (Adams et al., 2002) were used to prepare and analyze the anomalous signal from the scaled intensity data. Substructure solution, phasing and model building were carried out for different data sets using AutoSol and AutoBuild in PHENIX. Coot (Emsley \& Cowtan, 2004) was used to build missing residues and side chains into the autobuilt models. REFMAC5 (Murshudov et al., 1997) and phenix.refine (Afonine et al., 2005) were used for refinement. The theoretical Bijvoet ratio was calculated using the modified Hendrickson formula (Hendrickson \& Teeter, 1981; Dauter et al., 2002) and the measurability was calculated using phenix.reflections_statistics (Zwart et al., 2005). The quality of all atomic models was assessed with PROCHECK (Laskowski et al., 1993) and MolProbity (Chen et al., 2010). The figures were generated using Chimera (Pettersen et al., 2004) and PyMOL (http://www.pymol.org).
anomalous signal $(\langle\Delta F\rangle /\langle F\rangle)$ was 0.04 and was higher than the expected value $(0.009)$ as calculated using the Hendrickson formula (Hendrickson \& Teeter, 1981), $\left\langle\Delta F^{ \pm}\right\rangle /\langle F\rangle_{\text {calc }}=2^{1 / 2}$ $\times\left(f_{\mathrm{A}}^{\prime \prime} N_{\mathrm{A}}^{1 / 2}\right) /\left[f_{\text {eff }}(\theta) N_{\mathrm{P}}^{1 / 2}\right]$, where $N_{\mathrm{A}}$ and $f^{\prime \prime}$ are the number and imaginary scattering contribution of the anomalous scatterer, $N_{\mathrm{P}}$ is the total number of non-H protein atoms and $f_{\text {eff }}(\theta)$ is the average number of electrons for protein atoms. Apart from nine sulfurs, the numbers and atom types of extra bound anomalous scatterers were initially unknown. The structure was determined by the sulfur-SAD technique using the PHENIX package (Adams et al., 2002). Initially, seven sites were located and these sites were used for phasing. After phasing, a total of 18 sites were refined (Fig. $1 a$ and Table 2). At this stage, a preliminary model was built automatically and the obtained model was fed into AutoBuild for iterative model building and refinement. A total of 211 resides were built and 190 side chains were placed with $R_{\text {work }}$ and $R_{\text {free }}$ values of 0.21 and 0.23 , respectively. The amino-acid sequence information deduced from the cDNA was verified and modified based on the electron density at $1.5 \AA$ resolution. Preliminary structure analysis was carried out to identify the unknown bound anomalous scatterers (there was a choice between chloride and cadmium) since the saru-actinidin crystals were grown in $30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and cryosoaked with mother liquor supplemented with $20 \%$ glycerol. During substructure solution it was clear that the top six peaks belonged to unknown scatterers

Table 3
Summary of refinement statistics.

|  | $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cu} 300 \mathrm{~m} \mathrm{MCdCl}_{2}$ | $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ | $\mathrm{Cr} 300 \mathrm{~m} \mathrm{MCdCl}_{2}$ |
| :---: | :---: | :---: | :---: | :---: |
| Refinement |  |  |  |  |
| Resolution ( $\AA$ ) | 19.0-1.50 | 21.0-1.90 | 32.0-2.20 | 35.0-2.20 |
| No. of reflections in work set/test set | 29524/1992 | 13496/1501 | 9164/1018 | 9190/1022 |
| $R_{\text {work }} / R_{\text {free }}$ (\%) | 18.64/20.86 | 16.42/21.23 | 17.12/23.92 | 16.84/22.21 |
| No. of residues/waters | 220/288 | 220/216 | 220/153 | 220/141 |
| Cd ions | 5 | 8 | 5 | 7 |
| Stereochemistry |  |  |  |  |
| Bond lengths (A) | 0.001 | 0.007 | 0.010 | 0.009 |
| Bond angles ( ${ }^{\circ}$ ) | 0.887 | 0.983 | 1.066 | 1.032 |
| Ramachandran plot, residues in (\%) |  |  |  |  |
| Most favoured regions | 89.5 | 87.3 | 88.6 | 87.8 |
| Additionally allowed regions | 10.5 | 12.7 | 11.4 | 12.2 |
| Mean $B$ factors ( $\AA^{2}$ ) |  |  |  |  |
| Protein atoms | 15.1 | 19.0 | 22.0 | 18.2 |
| Cd ions | 19.9 | 32.3 | 36.8 | 27.7 |
| Water molecules | 28.9 | 32.8 | 27.8 | 23.1 |
| MolProbity |  |  |  |  |
| MolProbity score | 1.42 | 1.46 | 1.57 | 1.51 |
| All-atom clash score | 3.08 | 3.99 | 4.61 | 4.61 |
| Bad rotomers (\%) | 1.7 (3/174) | 2.3 (4/174) | 2.3 (4/174) | 2.3 (4/174) |
| Ramachandran favoured/outliers (\%) | 97.2/0.0 | 98.6/0.0 | 97.7/0.0 | 98.6/0.0 |


(b)

(c)

(d)

Figure 3
Overall structure of saru-actinidin. (a) Ribbon diagram of saru-actinidin. Two domains (L and R) and a hinge region are represented by blue, purple and orange colours, respectively. $(b, c)$ The modified activesite Cys25 in the cysteine sulfinic acid and cysteine sulfenic forms is shown together with difference Fourier ( $F_{\mathrm{o}}-F_{\mathrm{c}}$, green) and final ( $2 F_{\mathrm{o}}-F_{\mathrm{c}}$, blue) maps contoured at the $3.0 \sigma$ and $1.5 \sigma$ levels, respectively. (d) Superposition of the $\mathrm{C}^{\alpha}$ atoms of saru-actinidin from A. arguta Planch. and chins-actinidin from $A$. chinensis (blue), cysteine proteases (papain, caricain, chymopain and glycyl endopeptidase) from C. papaya (orange), ervatamin A, B and C from T. divaricata (purple), ginger protease II from Z. officinale (green), mexicain from J. mexicana (brown) and EP-B2 from H. vulgare (yellow). Loop regions with deviations are indicated as follows: $A$ (residues 59-62), $B$ (residues 100-104), $C$ (residues 172-175) and $D$ (residues 199201).
and most of these were on the protein surface. The interacting residues for these were Asp and Glu with distances between 2.0 and $2.5 \AA$. These parameters allowed us to assign the bound ions as cadmiums. In order to compare cadmium sites with sulfurs and cadmium/sulfur anomalous signals, we collected three further data sets (Cu300m $M \mathrm{CdCl}_{2}, \mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ ) from a further three crystals. All of these structures were also determined using the cadmium/sulfur-SAD technique. Initially, AutoSol located eight, five and seven sites; after phasing, additional sites were located and refined (11, 16 and 32 sites for $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$, $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$, respectively). Only two sulfur substructures (Fig. 1b) were located in the $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data set collected to $1.9 \AA$ resolution with $5 \%$ anomalous signal (in the presence of eight bound cadmiums) and 12.8 -fold overall redundancy. All S atoms could be located in the $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets collected to $2.2 \AA$ resolution using $\mathrm{Cr} K \alpha$ radiation (Figs. $1 c$ and $1 d$ ). The multiplicity required for the latter data was lower since the $f^{\prime \prime}$ values of sulfur and cadmium are doubled when using $\mathrm{Cr} \mathrm{K} \alpha$ radiation $(\lambda=2.29 \AA)$. Substructure solution, phasing and modelbuilding details are listed in Table 2. In all cases the figure of merit (FOM) was greater than 0.4 and the Bayesian CC was greater than 40. More than $95 \%$ of the residues and $90 \%$ of the side chains were placed automatically. Model completion, addition of waters and occupancy refinement of bound cadmium ions were carried out using Coot (Emsley \& Cowtan, 2004) and REFMAC5 (Murshudov et al., 1997). The final refinement parameters are listed in Table 3. Anomalous difference Fourier maps were computed and have been superimposed on the
. LPDYVDWRSSGAVVVDIKDQGQCGSCWAFSTIAAVEGINKIATGDLISLS 49 Saru .LPSYVDWRSAGAVVDIKSQGECGGCWAFSAIATVEGINKITSGSLISLS 49 2ACT LPEHVDWRAKGAVIPLKNOGKCGSCWAFSTVTTVESINOIRTGNLISLS 49 3BCN LPEQIDWRKKGAVTPVKNOGSCGSCWAFSTVSTVESINQIRTGNLISLS 49 IIWD LPSFVDWRSKGAVNSIKNQKQCGSCWAFSAVAAVESINKIRTGQLISLS 49 100E IPEYVDWRQKGAVTPVKNQGSCGSCWAFSAVVTIEGIIKIRTGNLNEYS 49 1PPN LPESVDWRAKGAVTPVKHOGYCESCWAFSTVATVEGINKIKTGNLVELS 49 1GEC . LPENVDWRKKGAVTPVKHOGSCGSCWAFSAVATVEGINKIRTGKLVELS 49 IMEG YPQSIDWRAKGAVTPVKNQGACGSCWAFSTIATVEGINKIVTGNLLELS 49 IYAL .LPDSIDWRENGAVVPVKNQGGCGSCWAFSTVAAVEGINQIVTGDLISLS 49 1CQD YPESIDWREKGAVTPVKNONPCGSCWAFSTVATIEGINKIITGQLISLS 49 2BDZ DLPPSVDWRQKGAVTGVKDQGKCGSCWAFSTVVSVEGINAIRTGSLVSLS 50 2FO5 . IPEYVDWRQKGAVTPVKNQGSCGSCWAFSAVVTIEGIIKIRTGNLNQYS 49 9PAP . LPENVDWRKKGAVTPVRHQGSCGSCWAFSAVATVEGINKIRTGKLVELS 49 1PPO


## EQELVDCGRTQNTRGCDGGFMTDGFQFIINNGGINTEANYPYTAEEGQCN 99 Saru

 EQELIDCGRTQNTRGCDGGYITDGFQFIINDGGINTEENYPYTAQDGDCD 99 2ACT EQQLVDCSKK. .NHGCKGGYFDRAYQYIIANGGIDTEANYPYKAFQGPCR 97 1PPN EQELVDCDTA. .SHGCNGGWMNNAFQYIITNGGIDTQQNYPYSAVQGSCK 97 1GEC EQELVDCDKK. . NHGCLGGAFVFAYQYI INNGGIDTQANYPYKAVQGPCQ 97 1MEG EQELLDCDRR. .SYGCNGGYPWSALQLVAQY.GIHYRNTYPYEGVQRYCR 96 1YAL EQELVDCDLQ. . SYGCNRGYQSTSLQYVAQN. GIHLRAKYPYIAKQQTCR 96 1CQD EQELVDCERR. .SHGCKGGYPPYALEYVAKN. GIHLRSKYPYKAKQGTCR 96 2BDZ EQELVDCDKH. .SYGCKGGYQTTSLQYVANN. GVHTSKVYPYQAKQYKCR 962 FO5 EQQLVDCTTA. .NHGCRGGWMNPAFQFIVNNGGINSEETYPYRGQDGICN 97 3BCN EQELLDCERR..SHGCDGGYQTTSLQYVVDN.GVHTEREYPYEKKQGRCR 96 9PAP EQELIDCDTAD. NDGCQGGLMDNAFEYIKNNGGLITEAAYPYRAARGTCN 99 1PPO EQELLDCDRR. . SYGCNGGYPWSALQLVAQY. GIHYRNTYPYEGVQRYCR 96100 E EQELVDCERR. . SHGCKGGYPPYALEYVAKN. GIHLRSKYPYKAKQGTCR 96 1IWD EQELVDCERR. .SHGCKGGYPPYALEYVAKN. GIHLRSKYPYKAKQGTCR 96 1PCI **: *: **LDLQQ. . EKYVSIDTYENVPYNNEWALQTAVAYQPVSVALEAAGYNFQH 146 VALQD. . .QKYVTIDTYENVPYNNEWALQTAVTYQPVSVALDAAGDAFKQ 146 AA. . . . . . KKVVRIDGCKGVPQCNENALKNAVASQPSVVAIDASSKQFQH 141 PYR. . . . . LRVVSINGFQRVTRNNESALQSAVASQPVSVTVEAAGAPFQH 142 AA. . . . . . SKVVSIDGYNGVPFCNEXALKQAVAVQPSTVAIDASSAQFQQ 141 SREKG. . .PYAAKTDGVRQVQPYNEGALLYSIANQPVSVVLEAAGKDFQL 143 ANQVG. . . GPKVKTNGVGRVQSNNEGSLLNA IAHQPVSVVVESAGRDFQN 143 AKQVG. . .GPIVKTSGVGRVQPNNEGNLLNA IAKQPVSVVVESSKGRPFQL 143 ATDKP . . . GPKVKITGYKRVPSNXETSFLGALANQPLSVLVEAGGKPFQL 143 STVN. . . .APVVSIDSYENVPSHNEQSLQKAVANQPVSVTMDAAGRDFQL 143 AKDKK. . .GPKVYITGYKYVPANDEISLIQAIANQPVSVVTDSRGRGFQF 143 VARAAQNSPVVVHIDGHQDVPANSEEDLARAVANQPVSVAVEASGKAFMF 149 SREKG. . .PYAAKTDGVRQVQPYNQGALLYSIANQPVSVVLQAAGKDFQL 143 AKQVG. . .GPIVKTSGVGRVQPNNEGNLLNAIAKQPVSVVVESKGRPFQL 143 AKQVG. . .GPIVKTSGVGRVQPNNEGNLLNA IAKQPVSVVVESKGRPFQL 143

YSSGIFTGPCGTAVDHAVTIVGYGTEG. GIDYWIVKNSWGTTWGEEGYMR 195 YASGIFTGPCGTAVDHAIVIVGYGTEG. GVDYWIVKNSWDTTWGEEGYMR 195 YKGGIFTGPCGTKLNHGVVIVGYGK. . . . . DYWIVRNSWGRHWGEQGYTR 186 YSSGIFTGPCGTAQNHGVVIVGYGTQS. GKNYWIVRNSWGQNWGNQGYIW 191 YSSGIFSGPCGTKLNHGVTIVGYQA..... NYWIVRNSWGRYWGEKGYIR 186 YRGGIFVGPCGNKVDHAVAAVGYGP . . . . . NYILIKNSWGTGWGENGYIR 188 YKGGIFEGSCGTKVDHAVTAVGYGKSG. GKGYILIKNSWGPGWGENGYIR 192 YKGGIFEGPCGTKVDHAVTAVGYGKSG. GKGYILIKNSWGTAWGEKGYIR 192 YKSGVFDGPCGTKLDHAVTAVGYGTSD. GKNYIIIKNSWGPNWGEKGYMR 192 YRSGIFTGSCNISANHALTVVGYGTEN. DKDFWIVKNSWGKNWGESGYIR 192 YKGGIYEGPCGTNTDHAVTAVGYGK. .... TYLLLKNSWGGNWGEEKGYIR 188 YSEGVFTGECGTELDHGVAVVGYGVAEDGKAYWTVKNSWGPSWGEQGYIR 199 YRGGIFVGPCGNKVDHAVAAVGYGP . . . . . NYILIKNSWGTGWGENGYIR 188 YKGGIFEGPCGTKVDHAVTAVGYGKSG. GKGYILIKNSWGTAWGEKGYIR 192 YKGGIFEGPCGTKVDGAVTAVGYGKSG. GKGYILIKNSWGTAWGEKGYIR 192

IQRNVG.GVGQCGIAKKASYPVKYYY 220
ILRNVG.GAGTCGIATMPSYPVKY. . 218
IKRGTGNSYGVCGLYTSSFYPVKKN. . 212
IRRASGNSPGVCGVYRSSYYPIKN. . 216
IKRAPGNSPGVCGLYKSSYYPTKN. . 216
LKRQSGNSQGTCGVYKSSYYPFKGFA 218
AERNIENPDGKCGITRFASYPVKK. . 216
IKRASGRSKGTCGVYTSSFFPIKG. . 212
VEKDSGASGGLCGIAMEASYPVKTY. 224
MKRVG. .GCGLCGIARLPFYPTKAX. 209
IKRGTGNSYGVCGLYTSSFYPVKN. . 212
IKRAPGNSPGVCGLYKSSYYPTKN. . 216
MLR.VG.GCGLCGIARLPYYPTKA. . 208
MERNVASSAGLCGIAQLPSYPTKA. . 215
IKRAPGNSPGVCGLYKSSYYPTKN . 216

* ** . * *
(a)

(b)

Figure 4
(a) Structure-based sequence alignment of some plant cysteine proteases. The proteins are saru-actinidin from A. arguta Planch. and chins-actinidin from A. chinensis (PDB code 2act; Baker \& Dodson, 1980), papain (1ppo, 9pap and 1ppn; Pickersgill et al., 1991, 1992; Kamphuis et al., 1984), caricain (1meg and 1pci; Katerelos et al., 1996; Groves et al., 1996), chymopain (1yal; Maes et al., 1996) and glycyl endopeptidase (1gec; O’Hara et al., 1995) from C. papaya (orange), ervatamin A (3ben; Ghosh et al., 2008), ervatamin B (1iwd; Biswas et al., 2003) and ervatamin C (1o0e; Thakurta et al., 2004) from T. divaricata, ginger protease II from Z. officinale (1cqd; Choi et al., 1999), mexicain from J. mexicana (2bdz; Gavira et al., 2007) and EP-B2 from H. vulgare (2fo5; Bethune et al., 2006). Identical/well conserved, conserved and semi-conserved residues are marked with asterisks, semicolons and dots, respectively. Active-site residues (red), L-domain and R -domain interface interaction residues (green) and cadmium-interacting residues (blue) are highlighted. (b) Residue conservation in plant cysteine proteases is shown as a schematic diagram. Identical, conserved, semi-conserved and weakly conserved residues are rendered in red, pink, grey and blue, respectively.

Table 4
Substructure solution from AutoSol (PHENIX), with each peak annotated with the corresponding atom.

| $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ |  | $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ |  | $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ |  | $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Site | Atom | Site | Atom | Site | Atom | Site | Atom |
| 1 | Cd1 | 1 | Cd1 | 1 | Cd1 | 1 | Cd1 |
| 2 | $\mathrm{Cd} 2 \dagger$ | 2 | Cd2 | 2 | Cd2 | 2 | Cd2 |
| 3 | Cd3 | 3 | Cd3 | 3 | Cd3 | 3 | Cd3 |
| 4 | Cd4 $\ddagger$ | 4 | Cd4 $\ddagger$ | 4 | Cd4 $\ddagger$ | 4 | Cd4 $\ddagger$ |
| 5 | Cd5 | 5 | Cd5 | 5 | Cd4\# | 5 | Cd5 |
| 6 | Cd4 $\ddagger$ | 6 | Cd6 | 6 | Cd5 | 6 | Cd4 $\ddagger$ |
| 7 | Met194 | 7 | Cd7 | 7 | Met194 | 7 | Cd6 |
| 8 | Met70 | 8 | Cd8 | 8 | Met70 | 8 | Cd7 |
| 9 | Cys156§ | 9 | Met194 | 11 | Cys156§ | 9 | Met194 |
| 10 | Cys 206§ | 10 | Cd4 $\ddagger$ | 12 | Cys229 | 10 | Met70 |
| 11 | Cys 229 | 11 | Cys22-Cys65† $\dagger$ | 13 | Cys206§ | 11 | Cys 22 ब |
| 12 | Cys56+† |  |  | 14 | Cys25 | 14 | Cys206§ |
| 13 | Cys $25 \S \S$ |  |  | 15 | Cys659 | 15 | Cys156§ |
| 14 | Cys659 |  |  | 16 | Cys56-Csy98† $\dagger$ | 16 | Cys659 |
| 15 | Cys $98 \pm \ddagger$ |  |  |  |  | 17 | Cys56+\# |
|  |  |  |  |  |  | 18 | Cys25§§ |
|  |  |  |  |  |  | 20 | Cys98\% $\ddagger$ |

$\dagger$ Alternate positions were refined. $\ddagger$ Two closely located cadmium sites. § Disulfide linkage Cys156-Cys206. © Disulfide linkage Cys22-Cys65. $\dagger \dagger$ For this disulfide bond, a super-sulfur peak was located. 俦 Disulfide linkage Cys56Cys98. §§ Free Cys.

Table 5
Sequence identity and r.m.s.d. of saru-actinidin with other known plant cysteine protease structures.

| PDB code | Identity (\%) | R.m.s.d. $(\AA)$ |
| :--- | :--- | :--- |
| 2act | 81.19 | 0.39 |
| 1ppn | 49.53 | 0.81 |
| 1gec | 51.39 | 0.79 |
| 1meg | 51.85 | 0.85 |
| 1yal | 53.21 | 0.85 |
| 1cqd | 62.50 | 0.62 |
| 2bdz | 51.42 | 0.80 |
| 2fo5 | 56.82 | 0.78 |
| 3bcn | 58.37 | 0.75 |
| 9pap | 49.06 | 0.79 |
| 1ppo | 52.31 | 0.87 |
| 1o0e | 60.10 | 0.70 |
| 1iwd | 62.79 | 0.58 |
| 1pci | 51.85 | 0.90 |

final atomic model to verify the assigned cadmium sites (Fig. 1).

### 3.2. Analysis of anomalous signals

The X-ray diffraction data were collected in a standard manner, in which the crystals were cryomounted in an arbitrary orientation and data sets were collected without using the inverse-beam technique. The expected anomalous signal $\left(\left\langle\Delta F^{ \pm}\right\rangle /\langle F\rangle\right)$ was calculated using the modified Hendrickson formula (Hendrickson \& Teeter, 1981; Dauter et al., 2002), $\left\langle\Delta F^{ \pm}\right\rangle\left\langle\langle F\rangle_{\text {calc }}=2^{1 / 2}\left[\left(f_{1 A}^{\prime \prime} N_{1 \mathrm{~A}}^{1 / 2}\right)+\left(f_{2 A}^{\prime \prime} N_{2 \mathrm{~A}}^{1 / 2}\right)\right] /\left[f_{\text {eff }}(\theta) N_{\mathrm{P}}^{1 / 2}\right]\right.$, where $f_{1}^{\prime \prime}$ and $f_{2}^{\prime \prime}$ are the imaginary scattering contributions of sulfur and cadmium ions, respectively, and $N_{1 \mathrm{~A}}, N_{2 \mathrm{~A}}$ and $N_{\mathrm{P}}$ are the number of sulfurs and cadmium ions and the total number of non- H atoms in the molecule, respectively. The average number of electrons for protein atoms $\left[f_{\text {eff }}(\theta)\right]$ is 6.7 . The present saru-actinidin molecule contains nine sulfurs, six cadmium ions and 1690 non-H atoms. The expected value of
the anomalous signal $\left(\left\langle\Delta F^{ \pm}\right\rangle /\right.$ $\langle F\rangle$ ) is about 0.07 and 0.13 for the $\mathrm{Cu} K \alpha$ and $\mathrm{Cr} K \alpha$ wavelengths, respectively. The observed Bijvoet ratio, measurability and anomalous signal-to-noise ratio in each data set are illustrated in Fig. 2. Generally, the experimentally measured Bijvoet ratios agree with the theoretical values in the low-resolution bins and the deviation increases with increasing resolution, probably owing to a decrease in the measurement accuracy (Dauter $e t$ al., 2002). In the cases of the $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}, \mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets the observed Bijvoet ratio mostly agrees with the calculated values except in the low-resolution bins (Fig. 2a) and this indicates that there were additional heavy atoms (i.e. cadmiums) in these crystals. This is so in the cases of the $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets, in which two or additional cadmium sites were located (Figs. $1 b$ and $1 d$ ). There were no additional sites in the case of the $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ data set; however, the located sites had high occupancy values compared with other data sets. In the case of the $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets the Bijvoet ratio is significantly higher in the highresolution bins and this is a consequence of reduced accuracy in the estimation of the high-resolution reflection intensities [i.e. relatively low $\langle I / \sigma(I)\rangle$ for high-resolution reflections; Table 1]. The increment from $2.7 \AA$ resolution onwards in the observed Bijvoet ratio is a consequence of poorer quality data in these data sets. In contrast to the anomalous signal, a sudden drop in measurability $|\Delta I| / \sigma(\Delta I)$ and anomalous signal-to-noise ratio from $2.7 \AA$ resolution onwards was observed for the $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets (Figs. $2 b$ and $2 c$ ). The mean anomalous signal-to-noise ratio was significant only to $<2.7 \AA$ resolution for the $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets (Fig. 2c), which suggested that $2.7 \AA$ resolution is an appropriate cutoff choice for substructure solution.

### 3.3. Comparison of anomalous scatterers

In order to verify the located atom types in the substruc-ture-solution peaks, we superimposed solution peaks on the corresponding final protein models. In all cases, the initially located sites are cadmium ions and were most likely to have originated from the crystallization solution (Fig. 1, Table 4). To verify that all anomalous scatterer sites had been located, an anomalous difference map was calculated using phases from the refined model, but no strong peaks were found. Of the top six peaks (including two sites located in proximity to each other; CD4 in Fig. 1), the first three sites showed high occu-
pancy. The occupancies (0.5-0.6) of other cadmium sites are in the same range as that of the Met194 S atom. In substructure solutions, only the Met194 S atom and Cys22-Cys65 (single peak, super-sulfur; Table 4 and Fig. 1b) were located for the $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data set, whereas in the other three data sets all sulfurs (Met and Cys) could be located (Table 4, Fig. 1). The cadmium ions and Met and free Cys sulfurs showed spherical densities, whereas the disulfide bond showed elliptical density (Fig. 1).

### 3.4. Description of the protease structure

Saru-actinidin has a papain-like fold and consists of two domains ( L and R ). The L domain is mostly $\alpha$-helical, while the R domain is built around a twisted $\beta$-sheet (Fig. 3). The $L$ domain consists of residues 14-111 and 217-220 and the R domain contains residues $1-9$ and $116-212$. There are three portions of the polypeptide chain (residues 10-13, 112-115 and 213-216) that act as linkers between the two domains and both the amino- and carboxyl-terminal ends of the polypeptide cross over to the other domain. The catalytic dyad residues Cys 25 ( L domain) and His162 ( R domain) are situated on either side of the inter-domain cleft. The active-site Cys 25 exists in the cysteine sulfenic acid form in the $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cr} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets, but in the cysteine sulfinic acid form in the $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ data set (Figs. $3 b$ and $3 c$ ).

### 3.5. Sequence and structural comparison with known plant cysteine proteases

The structure of saru-actinidin has been superimposed on other plant cysteine protease structures such as those of chinsactinidin, various cysteine proteases (papain, caricain, chymopain and glycyl endopeptidase) from C. papaya, ervatamin $\mathrm{A}, \mathrm{B}$ and C from T. divaricata, GPII from $Z$. officinale, mexicain from J. mexicana and EP-B2 from H. vulgare. The low r.m.s.d. values for $\mathrm{C}^{\alpha}$ atoms ( $<0.9 \AA$; Table 5) indicate a high similarity in overall conformation. Deviations are mainly observed in four surface-loop regions, which are designated regions $A-D$ in Fig. $3(d)$. The saru-actinidin primary sequence proposed here compares well with the amino-acid sequence of plant cysteine proteases, as illustrated in Fig. 4. The sequence identity of saru-actinidin is $81 \%$ to chins-actinidin, $63 \%$ to ginger

Figure 5

(a)

(b)

(c)

Polygonal water structures. (a) View of polygonal arrangements of hydration water molecules around the protein surface. Close-up stereoviews showing (b) a pentagonal ring, (c) fused tetragonal and pentagonal rings.

(e)

Figure 5 (continued)
Polygonal water structures. (d) Two fused pentagonal rings and (e) fused tetragonal and pentagonal rings forming a polygonal structure. Tetragonal and pentagonal rings are found suspended above the hydrophobic portions of side chains and are anchored by hydrogenbonding interactions with the surrounding polar atoms and water molecules (not shown). The reference and neighbouring molecules are shown in blue or pink/violet, respectively.
gonal rings of water molecules have previously been identified on the surface of proteins (Teeter, 1984; Kumaraswamy et al., 1996; Nakasako, 2004; Narayana, 2006; Britton et al., 2006). Water pentagons have also been found in the crystal structures of DNA fragments (Neidle et al., 1980). It is interesting to note that in the high-resolution saru-actinidin structure water molecules form a network of hydrogen bonds that resemble fused pentagonal rings (Fig. 5). A total of 13 such rings have been identified at six distinct places on the protease surface and most of these are constructed around a hydrophobic residue (Figs. $5 b-5 e$ ). The average interaction distance between these arranged waters is $\sim 2.8 \AA$ and the mean $B$ value is $\sim 26 \AA^{2}$ (which is lower than the value of $\sim 29 \AA^{2}$ for all waters).

### 3.7. Cadmium-binding sites

The residues involved in cadmium coordination are generally not conserved amongst plant cysteine proteases, indicating that the analyzed enzymes may not be cadmium-dependent proteins (Fig. 4a). Analysis of the four data sets that we have collected indicates that five cadmium sites are present in all data sets (Fig. $6 a$ and Table 4). The average coordination distance between cadmium ions and protein atoms and water molecules is $\sim 2.4 \AA$ and this value is comparable with previously reported values (Naismith et al., 1993; Dokmanić et al., 2008) and those for small molecules from the Cambridge Structural Database (CSD; Allen \& Kennard, 1993). Like zinc, cadmium possesses a filled $d^{10}$ orbital shell and is therefore able to accommodate different coordination geometries of similar energy (Pye et al., 2006; Dokmanić et al., 2008). The active-sitebound CD1 (cadmium site 1) displays distorted octahedral geometry and its coordination ligands are Cys25, His162, \#Asn220 and waters W1 and W2 (where \# indicates a
actinidin (Figs. $2 d$ and $3 a$ ). This insertion does not affect the overall fold and is not involved in active-site formation. However, it is located near the bottom of the active-site cavity.

### 3.6. Pentagonal arrays of water molecules

Water molecules play an important role in maintaining the structural stability of proteins and can control protein folding, structure and overall activity (Chaplin, 2001, 2006). Penta-
symmetry-related molecule). The sulfenic/sulfinic acid O atom, water molecules W 1 and W 2 and \#OXT of the C-terminal Asn220 form an octahedral base plane, whereas His $162 \mathrm{~N}^{\delta 1}$ and \#Asn220 O are located at the vertices of the octahedron (Figs. $6 b$ and $6 c$ ). The distance between CD1 and the active-site $\mathrm{S}^{\gamma}$ atom of Cys 25 is $\sim 3.2$ and $3.5 \AA$ for the $\mathrm{Cu} 30 \mathrm{~m} M \mathrm{CdCl}_{2}$ and $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data sets, respectively. The remaining cadmium sites are located on the protein surface, where they are involved in less elaborate inter-

(a)

(b)

(c)
(d)

(e)
$(f)$

(g)

(h)

(i)

(j)

Figure 6
Cadmium sites. (a) Cd ions in the protein involved in various coordination geometries such as $(b, c)$ distorted octahedral, ( $d$ ) distorted octahedral or square pyramidal, $(e)$ trigonal bipyramidal, $(f)$ square planar, $(g)$ trigonal, $(h, i)$ distorted square pyramidal and $(j)$ tetrahedral. The reference and symmetry-related molecules are shown in blue and purple, respectively. Bound cadmiums and waters are shown as cyan and red spheres, respectively. The marked coordination distances are in $\AA$.
molecular bridging and generally display a variety of coordinations (Figs. $6 d-6 j$ ). The carboxylate O atoms of Glu114, Glu191 and Asp111 act as symmetrical bidentate ligands for CD2, CD3 and CD8, respectively.

For sulfur-SAD phasing using a $\mathrm{Cu} K \alpha$ radiation source, the redundancy ( $>\sim 20$ ) and the resolution ( $<2.0 \AA$ ) can be very important factors in substructure solution. However, additional anomalous signal enhancers such as solvent $\mathrm{Cl}, \mathrm{P}$ and
naturally bound metals can contribute and allow easier substructure solution. This can be seen in the $\mathrm{Cu} 300 \mathrm{~m} M \mathrm{CdCl}_{2}$ data we collected, in which two sulfurs were located together with cadmium sites. In the data sets collected using $\mathrm{Cr} \mathrm{K} \alpha$ radiation all the S atoms were located, revealing that access to an in-house source of chromium-generated X-ray radiation will make it easier to solve protein structures using bound lighter atoms and metals. We propose that a dual source of

X-rays at in-house facilities coupled with the spiking of crystallization screens with atoms that are capable of anomalous scattering will be of considerable benefit.

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